Short Reports 317

pentayne (1) MS m/e: 162 M⁺, UV λ_{max}^{1sopentane} nm: 251.5 (69), 256.5 (79), 264.0 (100), 270.0 (86), 285.5 (84), 327.5 (1.5), 351.0 (2), 377.0 (2) and 409 (1.5). 11Z-Trideca-1.11-diene-3,5,7,9-tetrayne (2a). MS m/e: 164M⁺. UV $\lambda_{\text{max}}^{\text{isopentane}}$ nm: 257.5 (67), 270.0 (100). 287.0 (77), 314.5 (6), 336.0 (11), 361.0 (12) and 390.0 (8).11E-Trideca-1,11diene-3,5,7,9-tetrayne (2b). MS $m/e.164M^+$. $UV\lambda_{max}^{1sopentane}$ nm: 257.5 (73), 270.0 (100), 286.5 (75), 314.5 (6), 336.0 (10), 361.0 (12) and 390.0 (7). 3E-Trideca-1,3-diene-5,7,9,11-tetrayne (3). MS m/e. 164M⁺. UVλ_{max} nm: 262.5 (35), 272.5 (56), 288.5 (100), 310.0 (10), 330.5(15), 354.0(18) and 382.0(11). 3Z,11Z-Trideca-1,3,11-triene-5,7,9-triyne (4a). MS m/e: 166 $^+$. UV $\lambda_{mex}^{heptane}$ nm: 241.5 (51), 272.0 (100), 288.0 (95), 300.0 (23), 309.0 (16), 319.5 (25), 329.0 (18), 342.0 (52), 352.0 (13) and 368 (38). 3Z,11E-Trideca-1,3,11-triene-5,7,9triyne (4b). MS m/e: 166M⁺, UV $\lambda_{\text{max}}^{\text{heptane}}$ nm: 242.5 (52), 273.0 (100), 288.0 (93), 299.5 (24), 309.0 (18), 319.5 (39), 329.0 (19), 342.0 (54), 352.0 (13) and 367.5 (40). 3E,11E-Trideca-1,3,11-triene-5,7,9triyne (4c). MS m/e: 166M⁺, UV $\lambda_{\text{max}}^{\text{heptane}}$ nm: 243.0 (45), 273.0 (100), 288.5 (95), 300.0 (23), 310.0 (17), 320.0 (40), 329.5 (20), 343.0 (56), 353.0 (14) and 368.5 (41). Trideca-1,3,5-triene-7,9,11-triyene. A chromatographic fraction eluted after 4a-c and along with 5c showed a UV 283 nm peak [5]. Its MS, after subtraction of background due to 5c, shows m/e: 166M⁺ and a fragmentation pattern different from the mutually like patterns for 4a-c. 3E,5Z, 11E-Trideca-1,3,5,11-tetraene-7,9-diyne (5a). MS m/e 168M+ nm: 265.5 (78), 276.0 (60), 314.0 (76), 329.5 (100) and 354.0 (77). IR (CCl₄). 1623 and 1596 cm⁻¹ (C=C stretch), 1002 cm⁻¹ (—CH=CH₂, out-of-plane H deformation [0.0.p. H def.] of trans H's), 945 and 930 cm⁻¹ (trans —CH=CH o.o.p. \bar{H} def.), 904 cm⁻¹ (vmyl CH₂ wag) and 688 cm⁻¹ (probable cis—CH=CH—o.o.p. H def.). 3Z,5E,11E-Trideca-1,3,5,11tetraene-7,9-diyne (5b). MS m/e. $168M^+$. $UV\lambda_{max}^{heptane}$ 264.0 (61), 275.0 (51), 313.0 (70), 330.0 (100) and 354.5 (77). IR (CCl₄): 1623 and 1609 cm⁻¹ (C=C stretch), 968 cm⁻¹

(—CH=CH₂, o.o.p. H def of trans H's), 949 and 934 cm⁻¹ (trans —CH=CH—, o.o.p. H def.), 910 cm⁻¹(vinyl wag) and 648 cm⁻¹ (probable cis —CH=CH— o.o.p. H def.), 3E.5E.11E-Trideca-1,35,11-teraene-7,9-diyne (5c). MS m/e: 168M⁺. UV $\lambda_{\text{max}}^{\text{heptane}}$ nm: 263.0 (57), 274.5 (49), 313.0 (69), 330.0 (100) and 354.0 (79). IR (CCl₄): 1623 and 1600 cm⁻¹ (C=C stretch), 1002 cm⁻¹ (—CH=CH₂, o.o.p. H def. of trans H's), 965 and 947 cm⁻¹ (trans —CH=CH—, o.o.p. H def.) and 905 cm⁻¹ (vinyl wag).

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1-(3'-FURYL)-6,7-DIHYDROXY-4,8-DIMETHYLNONAN-1-ONE, A STRESS METABOLITE FROM SWEET POTATOES (*IPOMOEA BATATAS*)

LEO T. BURKA

Department of Chemistry, Vanderbilt University, Nashville, TE 37235, U.S.A.

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As part of our investigation of the stress metabolites of sweet potatoes we have isolated a new furanosesquiter-penoid from this source. The new compound (55 mg) was isolated from a MeOH-CHCl₃ extract of 2.1 kg of mercuric chloride treated sweet potato slices as an approximately 1:1 mixture with ipomeamaronol [1, 2] by Si gel chromatography (EtOAc-hexane, 1:2). Ipomeamaronol and compound 1 were separated by HPLC using a 60 cm C-18 μ Bondapak column (Waters Associates) eluted with MeOH-H₂O (1:1). The diol, $C_{15}H_{24}O_4$ (elemental analysis), crystallized from pentane-Et₂O, mp 70-71°, $[\alpha]_{16}^{26} + 17^{\circ}$ (c = 1.57, MeOH). The presence of the keto-furan moiety was

indicated by the UV, $\lambda_{max}^{\text{MeOH}} 252 \, \text{nm}$ ($\varepsilon = 2550$), and the IR $\lambda_{max}^{\text{CHCI}_3} 1675 \, \text{cm}^{-1}$. The broad absorption from 3700–3300 cm⁻¹ in the IR verified the presence of hydroxyl groups. The PMR and MS of the compound were consistent with the proposed structure: PMR (100 M Hz, CDCl₃); δ 0.94 and 0.97 (9 H, superimposed d, J = 7 Hz, Me's), 2.20 (2 H, br s, OH), 2.82 (2 H, t, J = 7 Hz, C-2), 3.14 (1 H, t, t = 5 Hz, C-7), 3.75 (1 H, t m, C-6), 6.84 (1 H, t m, 4-furyl), 7.56 (1 H, t m, 5-furyl), and 8.16 (1 H, t m, 2-furyl). The other protons in the molecule were accounted for in a multiplet from 1.1 to 2.1 (6 H): MS (probe) 70 eV t m/t (rel. int.); 268 [M⁺] (4), 250 [M⁺ - H₂O] (3), 232 [M⁺ - 2H₂O] (2%), 225 [M⁺ - C₃H₇] (3),

207 $[M^+ - C_3H_7 - H_2O]$ (4), 196 $[M^+ - C_4H_8O]$ (20), 195 $[M^+ - C_4H_9O]$ (27), 177 $[195 - H_2O]$ (22), 159 [cleavage at C-2, C-3] (6), 149 [cleavage at C-4, C-5] (22) 123 [cleavage of C-3, C-4] (59), 110 [McLafferty cleavage of C-2, C-3] (35), and 95 [cleavage of C-1, C-2] (100). Reaction of compound 1 with periodic acid followed by silver oxide oxidation gave the keto acid 2 (mp 46-47°, $[\alpha]_a^{25} = -12^\circ$) which was identical to that obtained from periodic acid oxidation of 7-hydroxymyoporone, 3 [3].

To gain further information about the relative configuration of the hydroxyl groups in 1, 100 mg of compound 3 was reduced with 1.9 eq of sodium bis(2methoxyethoxy) aluminum hydride. The reduction gave a 4:1 mixture of ketodiols which were separated by HPLC using a 60 cm C-18 μBondapak column eluted with MeOH-H₂O (1:1). The major product, mp 94-95°, $[\alpha]_d^{25} + 26^\circ$ was assigned the erythro configuration since reduction of ketols with hydride reagents has been shown to give predominantly the erythro isomer [4, 5]: PMR (100 MHz, CDCl₃) δ 0.89 and 1.02 (9 H, superimposed d, J = 7 Hz, Me's), 1.1-2.0 (6 H, m, C-3, C-4, C-5, C-8), 2.53 (2 H, br s, —OH), 2.82 (2 H, t, J = 7Hz, C-2), 3.35 (1 H, dd, J = 4 and 8 Hz, C-7), 3.82 (1 H, m, C-6), 6.83 (1 H, m, 4-furyl), 7.50 (1 H, m, 5-furyl), and 8.13 (1 H, m, 2-furyl). The minor product, mp 45-47°, $[\alpha]_d^{25}$ -17° , was assigned the three configuration: PMR (100 MHz, CDCl₃) δ 0.96 and 0.98 (9 H, superimposed d, J = 7 Hz, methyls, 1.1-2.1 (6 H, m, C-3, C-4, C-5,C-8), 2.84 (2 H, t, J = 7 Hz, C-2), 3.15 (1 H, t, J = 5 Hz, C-7), 3.84 (1 H.m, C-6), 6.82 (1 H, m, 4-furyl), 7.49 (1 H, m,

5-furyl), and 8.12 (1 H, m, 2-furyl). The IR and MS of both reduction products were indistinguishable from those of the natural product.

The PMR spectra of 1 most closely resembles the spectrum of the minor reduction product, especially the chemical shift and spin-spin coupling of the proton on C-7, and 1 is assigned the *threo* configuration at C-6 and C-7. We have shown that the keto-acid 2 from 3 is levorotatory [3], thus, the configuration at C-4 must be the same for all 3 keto-diols. The minor reduction product and 1 must have a mirror image relationship at C-6 and C-7. This relationship implies that 1 and 3 result from divergent biosynthetic pathways and that neither is the precursor of the other.

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